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Bilayered structure of N,N'-diphenyl-4,4'-biphthalimide

KENJI OKUYAMA, IGOR ROZHANSKII AND KOHEI GOTO

"Faculty of Technology, Tokyo University of Agriculture and Technology, Koganei, Tokyo 184-8588, Japan, and ^bTsukuba Research Laboratory, JSR Corporation, Miyukigaoka, Tsukuba, Ibaraki 305-0841, Japan. E-mail: okuyamak@cc. tuat.ac.jp

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Abstract

The molecular and crystal structure of the title compound ($C_{28}H_{16}N_2O_4$), which corresponds to the monomer unit of the thermally stable polyimide, has been determined. The molecule is composed of a central phthalimide plane with phenyl ring planes at both ends twisted by $61.24(8)^\circ$. The repeating unit along the c axis consists of a two-layer structure, in which the molecules tilt in opposite directions in adjacent layers.

Comment

Thermally stable polyimide, obtained by condensation polymerization of 3,3',4,4'-biphenyltetracarboxylic dianhydride and bis(4-aminophenyl) ether, is one of the commercially available polymers that have several industrial applications. The title compound, (I), was synthesized using aniline instead of bis(4-aminophenyl) ether to investigate the stereochemistry and the physical properties of the chemical repeating unit of this polyimide.

Compound (I) has an inversion center in its chemical structure which coincides with the crystallographic inversion center. The molecule consists of a phthalimide plane at the center, with phenyl rings at both ends. The phthalimide plane, which is defined by N1, C7, C8, C9, C10, C11, C12, C13 and C14, has good planarity with a maximum deviation of 0.027 (2) Å. The phenyl ring plane is twisted from the phthalimide plane by 61.24(8)°. The molecules are packed in a bilayered structure in which they tilt in opposite directions in adjacent layers. The carbonyl-O atoms have several short contacts with C atoms in adjacent molecules $[O1 \cdots C6 \ 3.234(3)]$ and $O2 \cdots C10 \ 3.299(2)$ Å]. The distances between the O atoms and the H atoms attached to C6 and C10 are shorter than the van der Waals contacts, which suggests C—H···O hydrogen bonding.

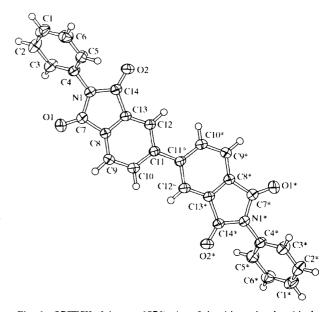


Fig. 1. ORTEPII (Johnson, 1976) plot of the title molecule with the atomic numbering scheme and non-H atoms represented by 50% probability displacement ellipsoids. Symmetrically generated atoms are denoted by *.

Experimental

The title compound was synthesized from 3,3',4,4'-biphenyltetracarboxylic dianhydride and aniline. Single crystals used for X-ray diffraction were colorless and prismatic, and were obtained by the slow sublimation of a powdered sample in 2 mmHg.

Crystal data

$C_{28}H_{16}N_2O_4$ $M_r = 444.45$ Monoclinic $P2_1/c$ a = 8.3987 (15) Å b = 6.890 (4) Å	Mo $K\alpha$ radiation $\lambda = 0.7107 \text{ Å}$ Cell parameters from 25 reflections $\theta = 21.9-24.8^{\circ}$ $\mu = 0.097 \text{ mm}^{-1}$
c = 17.8585 (15) Å	T = 296.2 K
$\beta = 90.868 (11)^{\circ}$	Prismatic
$V = 1033.3 (7) \text{ Å}^3$	$0.60 \times 0.40 \times 0.30 \text{ mm}$
Z = 2	Colorless
$D_x = 1.428 \text{ Mg m}^{-3}$	
$D_m = 1.425 \text{ Mg m}^{-3}$	
D_m measured by flotation in	
aqueous KI	

Data collection

Rigaku AFC-5R diffractom-	$R_{\rm int} = 0.016$
eter	$\theta_{\rm max} = 30.0^{\circ}$
ω –2 θ scans	$h = 0 \rightarrow 11$
Absorption correction: none	$k=0\rightarrow 9$
3608 measured reflections	$l = -24 \rightarrow 24$
3273 independent reflections	3 standard reflections
1833 reflections with	every 100 reflections
$I > 2\sigma(I)$	intensity decay: 0.14%

Refinement

Refinement on F	$(\Delta/\sigma)_{\text{max}} = 0.002$
R = 0.049	$\Delta \rho_{\text{max}} = 0.22 \text{ e Å}^{-3}$
wR = 0.067	$\Delta \rho_{\min} = -0.28 \text{ e Å}^{-3}$
S = 1.145	Extinction correction: none
1833 reflections	Scattering factors from
154 parameters	International Tables for
H-atom: see text	Crystallography (Vol. C)
$w = 1/[\sigma^2(F_o) + 0.0009 F_o ^2]$	

Table 1. Selected geometric parameters (Å, °)

	**	•	
O1—C7	1.196(2)	C5C6	1.385(3)
O2—C14	1.205 (2)	C7—C8	1.480(3)
N1—C4	1.434(2)	C8—C9	1.377(2)
NI—C7	1.409(2)	C8C13	1.384(3)
N1—C14	1.396(3)	C9—C10	1.384(3)
C1—C2	1.369(3)	C10C11	1.401(3)
C1—C6	1.378(3)	C11—C111	1.488(3)
C2C3	1.382(3)	C11C12	1.405(2)
C3—C4	1.376(3)	C12C13	1.371(3)
C4—C5	1.383(3)	C13—C14	1.486(2)
C4N1C7	123.7 (2)	C7—C8—C13	108.7 (2)
C4-N1-C14	124.4(2)	C9C8C13	120.6(2)
C7-N1-C14	111.8(1)	C8-C9-C10	117.4(2)
C2C1C6	120.3(2)	C9-C10-C11	123.0(2)
C1—C2—C3	120.6(2)	C10-C11-C11'	120.9 (2)
C2—C3—C4	119.0(2)	C10-C11-C12	118.3 (2)
N1—C4—C3	119.8 (2)	C11'C11C12	120.9(2)
N1C4C5	119.1 (2)	C11—C12—C13	118.3 (2)
C3—C4—C5	121.1(2)	C8—C13—C12	122.5 (2)
C4—C5—C6	119.1 (2)	C8-C13-C14	108.3(2)
C1C6C5	119.9(2)	C12-C13-C14	129.2 (2)
O1C7N1	125.0(2)	O2—C14—N1	125.5 (2)
O1—C7—C8	129.5 (2)	O2-C14-C13	128.7 (2)
N1—C7—C8	105.5(2)	N1—C14—C13	105.8(2)
C7—C8—C9	130.8 (2)		

Symmetry code: (i) 1 - x, -y, 1 - z.

a glass tube oven at about 543 K under reduced pressure at H atoms were fixed geometrically and were not included in the refinement procedure.

> Data collection: MSC/AFC Diffractometer Control Software (Molecular Structure Corporation, 1988). Cell refinement: MSC/AFC Diffractometer Control Software. Data reduction: TEXSAN (Molecular Structure Corporation, 1995). Program(s) used to solve structure: SHELXS86 (Sheldrick, 1985). Program(s) used to refine structure: TEXSAN. Software used to prepare material for publication: TEXSAN.

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> Supplementary data for this paper are available from the IUCr electronic archives (Reference: OA1075). Services for accessing these data are described at the back of the journal.

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Equilin

MARK W. SAWICKI, A NAIYIN LI AND DEBASHIS GHOSH A, B

"Roswell Park Cancer Institute, Elm and Carlton Streets, Buffalo, NY 14263, USA, and ^bHauptman-Woodward Medical Research Institute, 73 High St., Buffalo, NY 14203, USA. E-mail: ghosh@hwi.buffalo.edu

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Abstract

3-Hydroxyestra-1,3,5(10),7-tetraen-17-one, $C_{18}H_{20}O_2$, crystallizes in space group $P2_12_12_1$ from ethyl acetate. The planarity of the B ring, and the difference in puckering of the C and D rings from that of estrone, are due to the presence of the C7=C8 double bond, which may explain its function as an inhibitor of human type 1 17β -hydroxysteroid dehydrogenase, instead of being its substrate.